

Fabrication of Self-powered Heterostructure MoS₂/Cu₂O Photoelectrochemical Biosensor

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Abstract--Abstract- In this study, chemical vapor deposition (CVD) was used to grow the MoS₂ thin films, and electrochemical deposition was used to grow the Cu₂O thin films. We applied mechanical grinding to polishing the surface of the grown Cu₂O and made the first treatment of the grown MoS₂. Then we transferred MoS₂ to Cu₂O which had been polished to complete our heterostructure. The analysis of the transmission electron microscopy (TEM) is used to observe the crystal structure of the MoS₂ flakes. By observing periodicity diffraction points from the corresponding selected area diffraction pattern (SAD) on Cu₂O thin film indicates the monocrystalline with our grown sample. In addition, the measurements of Raman spectroscopy, Multiphoton excitation microscope, Atomic force microscope (AFM) and Photoluminescence (PL), we will able to confirm the monolayer structure of the MoS₂ flakes. After preliminary processing of the grown MoS₂ flakes, the sample transfer to a Cu₂O thin film to complete the p-n heterojunction structure. We analyzed the luminous energygap of the p-n heterojunction structure by using the measurements of Scanning electron microscope (SEM), Second-harmonic generation (SHG), and Raman mapping. It showed that Cu₂O thin film prepared by the deposition environment pH value 12 had a highly regular crystallinity and its crystal lattice is complete, while MoS₂ is a single-layer structure. Subsequently, we measured the biosensor through the UV-Visible spectrometer and the micro-current meter. Through the steps of the above process, we successfully manufactured a photoelectrochemical biosensor. We used it for photo-response measurement of two cancer cells with different stages of cancerization. Then we obtained the photo-response characteristics and the working wavelength range of our biosensor. In this cancer cell measurement experiments, the characteristic response of biochips to cancer cells and the different cancerous staging characteristics were verified.

keywords—Photoelectrochemical; Chemical Vapor Deposition (CVD); MoS₂; Cu₂O

Introduction

In recent years, two-dimensional (2D) semiconducting transition metal dichalcogenides (TMDs), including MX₂ (M = Mo, W; X = S, Se), have attracted a great deal of attention because of their unique structure as well as remarkable physical and chemical properties. Molybdenum disulfide (MoS₂), a typical TMDC with a direct band gap of 1.8 eV for the monolayer and a layer number dependent band structure, has been used to tackle the zero-bandgap problem of graphene. Mechanical exfoliation is widely used in fundamental research because of the possibility to fabricate high quality 2D materials. Bulk MoS₂ is an indirect band gap semiconductor with a band gap of 1.3 eV. The chemical vapor deposition (CVD) method is one of the most practical methods to prepare 2D materials, including graphene, boron nitride (BN), and MoS₂-like TMDCs.

The Cu₂O layers are prepared by several techniques such as the sputtering technique, the thermal oxidation of a metallic Cu sheet, anodic oxidation, photochemical deposition and electrodeposition in an aqueous solution containing copper sulfate hydrate and lactic acid. Cu₂O has a small band gap of ~2.0 eV and suitable conduction band, which gives it efficient visible light absorption. Furthermore, copper is naturally abundant, which makes for possible large-scale fabrication of p-type Cu₂O photoelectrodes, offering potential competitiveness over other semiconductors. The p-n heterogeneous structure was prone to photoelectrochemical reactions and self-powered.

Method

The results presented in this paper were reproduced more than five times, and the phenomenon of the domain shape change in the same place on the chip along the flow direction always existed. We used CVD to grow MoS₂ with one furnace to control the temperature of MoO₃ and S separately, and a ceramic boat to create a wider change in MoO₃ concentration on the substrates. Growth substrates were Si with a 300 nm layer of SiO₂. Substrates were cleaned in acetone for 10 min and then isopropyl alcohol for 10 min, followed by DI-water for 10 min. After being cleaned, four substrates were tightly aligned and placed face down above a ceramic boat containing 10 mg of molybdenum(VI) oxide (MoO₃) powder (10um 99.95%, Gredmann). Quartz tube together with another boat containing 1 g of sulfur powder ($\geq 99.98\%$, Gredmann). They were then put into furnace (furnace having MoO₃ and S is in far away). The CVD growth occurred at atmospheric pressure while ultra-high-purity argon was flowing. The CVD system was 100 sccm of Ar gas for 1.25 h when the temperatures was on 700°C, held at the setting temperature for 45 min.

The Cu₂O film was formed by electrochemical deposition. The electrolyte was made of 0.4M CuSO₄ and 85% lactic acid. pH was regulated by adjusting the concentration of NaOH. Three electrolytes with different pH were used to determine the photochemical response of the biosensor chip.

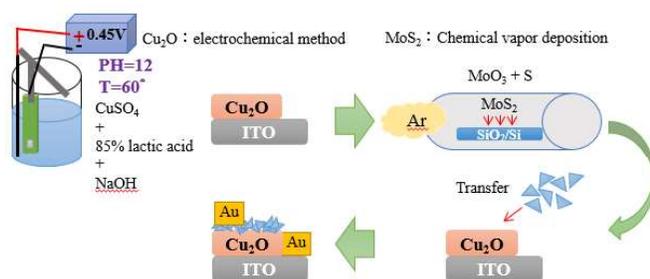


Fig. 1. Fabrication of Monolayer n-type MoS₂ grown by Chemical vapor deposition on p-type Cu₂O Photoelectrochemical.

Results and discussions

The SEM images of the p-Cu₂O with mechanical smooth is show in Fig.2. The SEM images of the p-Cu₂O with mechanical smooth is show in Fig.2. Measurements indicate that the Cu₂O thin films are mechanical panarization about 2 μ m and that the Cu₂O film thickness for AFM image Z range is 94.623nm and rms is 3.849nm show in Fig.3. The microstructure and properties of the MoS₂ crystals were measured by SEM, Raman spectra, AFM, and PL spectra, confirming that the MoS₂ film was a uniform, single layer with high crystallinity. In addition, the effects of MoO₃ precursor temperature and Ar gas flow rate on MoS₂ crystal shape were also investigated. Two characteristic Raman vibration modes can be seen in the spectra in Fig.4 the E₁g mode representing the in-plane vibration of molybdenum and sulfur atoms and the A₁g mode related to the out-of-

plane vibration of sulfur atoms. The fitting results show that these two modes are located at 384.7 and 405.0 cm⁻¹, respectively, giving a frequency difference Δk of 20.4 cm⁻¹. Most of the domain has a Δk of <20 cm⁻¹, confirming a homogeneous monolayer. Fig.5a shows the typical PL spectra from a synthesized MoS₂ domain. A strong PL signal is located at 675 nm. Panels b of Fig.4 show corresponding PL intensity maps for a large triangular MoS₂ domain.

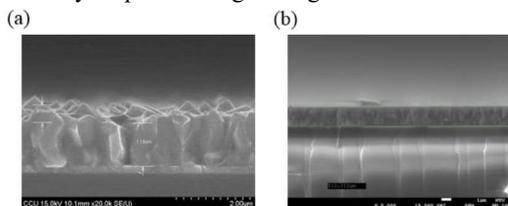


Fig. 2. SEM images of the p-Cu₂O/ITO, Mechanical smooth SEM of Cu₂O.

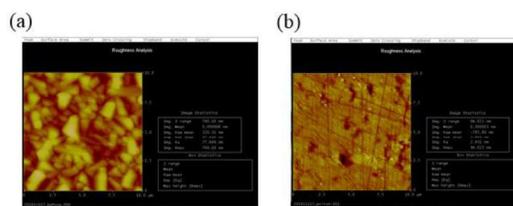


Fig. 3. AFM image for Cu₂O mechanical smooth, the Z range is about 94.623nm and the rms is 3.849nm.

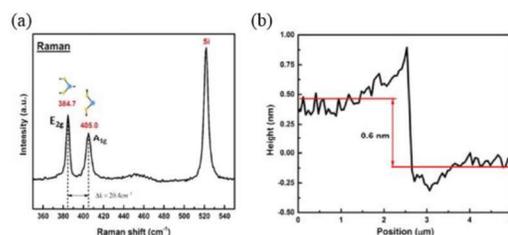


Fig. 4. (a) Raman spectrum of the MoS₂ domain, plotting the spatial variation of the magnitude of the frequency difference between A₁g and E₁g. (b) The height profile of AFM image for MoS₂ domains.¹

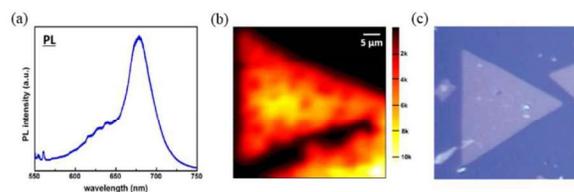


Fig. 5. (a) Photoluminescence (PL) spectrum of a synthesized MoS₂ domains. (b) 2D image of the PL intensity of triangular MoS₂ domains. The excitation wavelength is 532 nm. (c) The image of the optical microscope of monolayer MoS₂.¹

Conclusion

The Cu₂O could be combined with certain n-type semiconductors of the more positive conduction bands such as MoS₂, ZnO, rGO, etc., forming the n-p junction to allow the efficient transport of photogenerated electrons from Cu₂O to the n-type semiconductor conduction band that results in the improved photostability of Cu₂O. After MoS₂ grown on 300 nm SiO₂/Si substrate by CVD process, the MoS₂ flakes were transferred to Cu₂O thin film to finish the p-n heterogeneous structure. The p-n heterogeneous structure was prone to photoelectrochemical reactions and self-powered. We researched electrical properties variety and characteristic of photovoltaic without requiring an extra bias voltage. We hope the different biosensor of cancer level cells can be applied in the future.

Acknowledgment

This research was supported by the Ministry of Science and Technology, The Republic of China, under the Grants MOST 104-2221-E-194-054, 105-2923-E-194-003-MY3, and 105-2923-E-194-002-MY3.

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