

Growth of improved ultra-thin pristine SnO₂ films by electric field modified spray pyrolysis and DC reactive sputtering techniques

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ABSTRACT

The present paper reports on the deposition of high conductivity ($10^2 \text{ ohm}^{-1} \cdot \text{cm}^{-1}$) SnO₂ ultrathin films (less than 50 nm) by spray pyrolysis and pulsed dc magnetron reactive sputtering techniques in the presence of electric field. A comparative study has been carried out on the electrical, structural and optical properties of SnO₂ films deposited on a glass substrate with and without the presence of electric field. This is possibly the first such report of fabricating such high conductivity ultrathin films by spray pyrolysis. An evolution of crystallographic structure with increasing field is observed.

Keywords: ultrathin films spray pyrolysis, SnO₂, reactive sputtering, electric field.

1 INTRODUCTION

Device miniaturization and high density ICs demand high quality nanocrystalline transparent conducting oxide (TCO) films. These films show unique properties on the nanometre scale, and their extensive use in modern semiconductor device applications have prompted their widespread investigations [1, 2]. The physical properties of ultrathin films of these functional materials strongly depend on the crystallite size and structure/roughness of the interface. Pure SnO₂, an *n*-type semiconductor, also belongs to the class of TCOs. It has a rutile structure and a wide energy gap $\sim 3.6 \text{ eV}$ [3]. Owing to its outstanding electrical, optical, and electrochemical properties, SnO₂ is extensively used in many applications such as catalytic support material, transparent electrodes, touch screens, flat panel displays, solar cells, and gas sensors [4, 5, 6]. It has been reported that the electrical properties of the films strongly depend on the size, orientation, and shape of SnO₂ grains in films. Due to its potential applications in such devices that require films of thicknesses 1 – 50 nm, it is now important to have a re-look on the material properties through fundamental research [7], and hence calls for a systematic investigation on the microstructure, and interface of ultrathin films and their effect on various physical properties. There are several techniques that can be employed for the fabrication of different oxide thin films including reactive sputtering [8], evaporation [9], chemical vapour deposition [9, 10], dip coating [11] and spray pyrolysis [12, 13, 14]. Unlike physical vapour deposition techniques which require

vacuum of high order, spray pyrolysis is a simple and low cost deposition technique of preparing transparent and conducting oxide films of uniform thickness. We have initiated efforts to modify this simple technique for preparation of films < 50 nm thickness. In this paper we present some of our results on growth and properties of pristine ultrathin SnO₂ films, using electric field assisted spray pyrolysis and pulsed dc magnetron reactive sputtering techniques.

2 EXPERIMENTAL

2.1 Spray pyrolysis

A schematic diagram of the spray pyrolysis set up employed for the growth of pristine SnO₂ films is shown in figure 1. Undoped tin oxide films were fabricated on the glass substrates at a temperature of 450°C using 0.1M solution of SnCl₄.5H₂O in absolute ethanol. The flow rate (5 ml/min) of solution was controlled by a carrier gas at a pressure of 0.5 kg/cm². The nozzle was at a distance of 24 cm from the substrate during deposition. A potential difference of 32V was applied on the substrate during deposition.

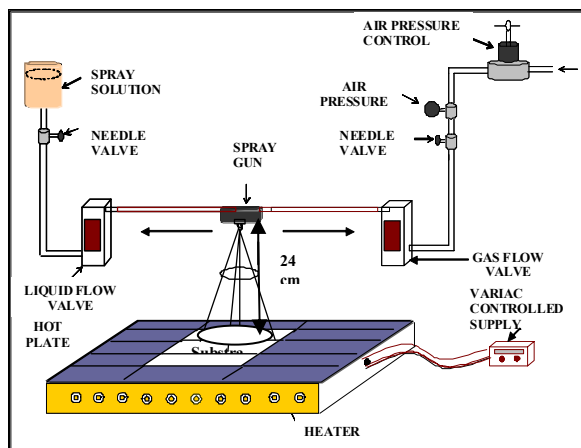


Fig. 1 Schematic diagram of the Spray pyrolysis system

2.2 DC Magnetron Reactive Sputtering

Tin oxide films were also prepared by pulsed dc magnetron reactive sputtering technique. The target was a metallic disk of Sn (purity 99.9%). The target–substrate distance was 10 cm. A thermocouple was positioned on the reverse side of the substrate holder to control the substrate temperature during the film growth, the sputtering gases were O₂ and Ar, which were combined in the sputtering chamber. Before deposition, the sputtering chamber was pumped down to 8 × 10⁻⁶ Torr. The target was pre-sputtered in O₂ for 15 min, in combination with a shutter, ensuring stabilized sputtering conditions. Total sputtering pressure of 2.2x10⁻² Torr was used. The Ar flow rate was 15 sccm and oxygen flow rate was maintained at 3.5 sccm. During deposition the substrate temperature was 350°C and the deposition rate was 1.5nm/min.

The film thickness was measured using XRR and ellipsometry. Optical transmission studies of all the films prepared by both the methods were made using a UV-VIS IR double beam spectrophotometer. The crystallographic structure of the films was studied by X'PertPRO X-ray diffraction system using CuK α radiation. The electrical properties were studied by Van der Pauw method.

3 RESULTS AND DISCUSSION

3.1 Structural Analysis

The XRD patterns obtained for the films grown on glass substrates were studied in the 2 θ range of 20°-80°. The XRD patterns (Fig. 2) show that the SnO₂ films are polycrystalline in nature irrespective of the technique used for deposition. The matching of the observed and the standard d-values from JCPDS data PDF #770449 confirms that the deposited films are of SnO₂ with tetragonal structure. All the patterns contain the characteristic SnO₂ peaks only. The presence of other phases such as SnO, Sn₂O₃ is not found in the present study. The pristine films grown by both the techniques are therefore single phase. The lattice constant 'a' and 'c', for the tetragonal phase structure is determined by the relation

$$\frac{1}{d^2} = \left(\frac{h^2 + k^2}{a^2} \right) + \left(\frac{l^2}{c^2} \right) \quad (1)$$

where 'd' is the interplaner distance and (hkl) are miller indices, respectively. The calculated average lattice parameter values are a = 4.7 Å and c = 3.2 Å.

The lattice parameter values are in good agreement with the standard data. The average grain size of the SnO₂ thin films was calculated by using Scherrer's formula [8].

$$D(nm) = \left(\frac{0.9\lambda}{\beta \cos\theta} \right) \quad (2)$$

' λ ' is wavelength of X-ray, ' β ' is the full-width at half of the peak maximum in radians and ' θ ' is Bragg's angle. The grain size of these samples is given in Table 1.

Table 1

Deposition Technique	Thickness (nm)	Grain size (nm)
Spray Pyrolysis Without Electric Field	30	24.5
Spray Pyrolysis With Electric Field of 32V/cm	30	18.6
Pulsed DC magnetron reactive sputtering without Electric Field	30	19.9

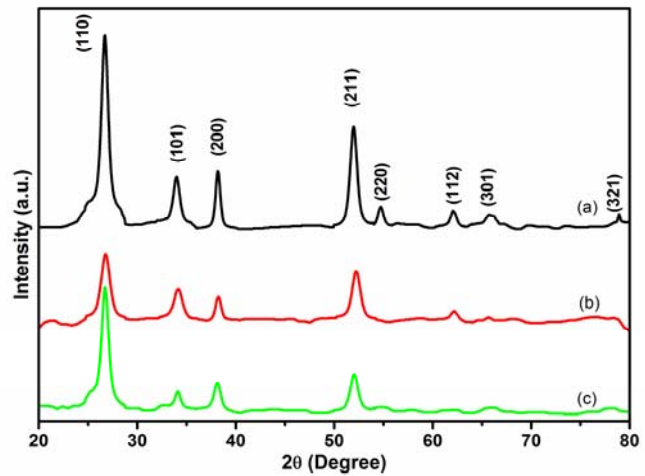


Fig.2. XRD pattern of SnO₂ films of thickness 30nm deposited by spray pyrolysis by applying (a) zero electric field and (b) 32V/cm electric field at the substrate during deposition, and (c) by pulsed dc magnetron reactive sputtering technique at zero electric field.

Table 2

Applied Electric Field (V/cm)	Grain size (nm)
0	19.9
14	20.2
32	21.3

Figure 3 shows the XRD pattern of ultrathin films deposited by pulsed dc magnetron reactive sputtering at different field values showing a trend of systematic prominence of (110) peak at the expense of (200) peak. Further study is in progress for a clear picture. The variation of the grain size with applied electric field is given in Table 2.

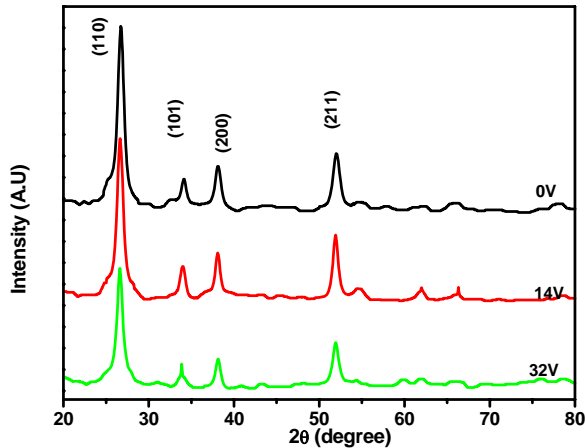


Fig.3. XRD pattern of SnO₂ films of thickness 30nm deposited by pulsed dc magnetron reactive sputtering technique by applying different electric field

3.2 Electrical Properties

The values of Hall mobility, carrier density and resistivity measured at room temperature are shown in Table 3. The values of μ , n and ρ for SnO₂ films deposited by different techniques are obtained from the combined measurements of resistivity and Hall coefficient. The electrical conductivity is increased by one order i.e. from 3.8 to 14.7ohm⁻¹cm⁻¹, SnO₂ films when films are deposited under electric field by spray pyrolysis technique. It is also found that the films deposited by sputtering technique are more conducting than the films deposited by spray pyrolysis. The decrease in the resistivity can be explained on the basis of the observed increase in the carrier concentration, from 7.04x10¹⁷ to 7.24x10¹⁸ cm⁻³, for the film deposited by spray pyrolysis in the presence of electric field.

Table 3

Deposition Technique	T _s (°K)	R _s (Kohm / sq)	N (per cc)	μ (cm ² / V. s)	ρ (Ω cm)
Spray Pyrolysis Without Electric Field	450	88.40	7.04x 10 ¹⁷	25.49	2.6x 10 ⁻¹
Spray Pyrolysis With Electric Field of 32V/cm	450	22.75	7.24x 10 ¹⁸	9.6	6.8x 10 ⁻²
Pulsed DC magnetron reactive sputtering	350	10.19	2.38x 10 ¹⁹	8.58	3.0x 10 ⁻²

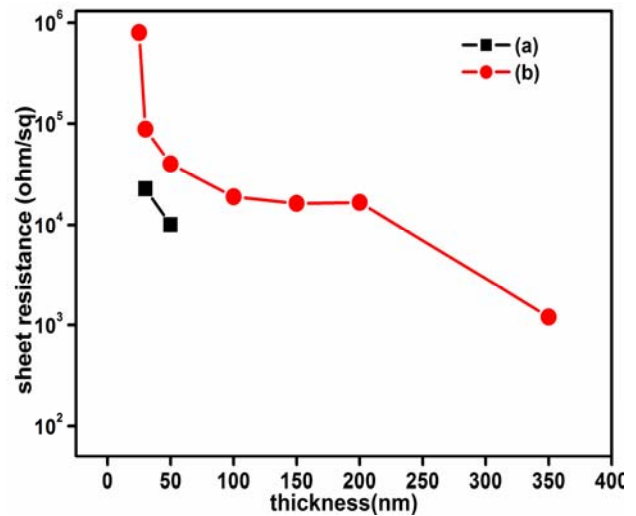


Fig.4. Sheet resistance vs. thickness plot for the spray pyrolysed SnO₂ films (a) with and (b) without electric field.

Figure 4 shows the variation of sheet resistance with the film thickness of the SnO₂ films deposited by spray pyrolysis without electric field and in the presence of electric field (32V/cm). We can observe () that for a film of thickness 30 nm the sheet resistance drops by about a factor of 40 if field is applied during deposition.

3.3 Optical properties

Figure 5 shows the transmission spectra of our spray pyrolysed and sputtered films. The films by spray pyrolysis are highly transparent (90%) in the visible region. However, films deposited at 350°C by sputtering are found to be some what less transparent in this region.

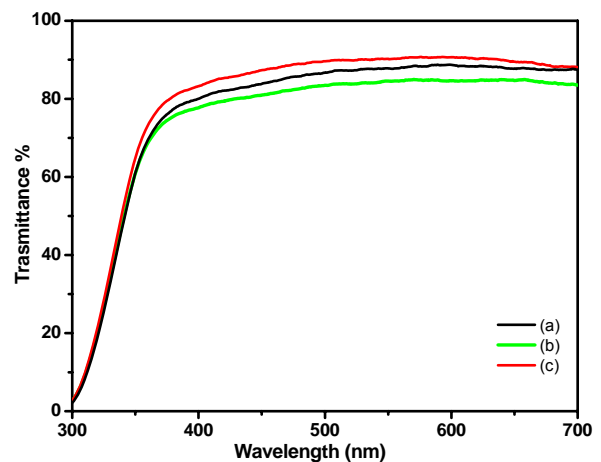


Fig.5 Optical transmittance spectra of SnO₂ films (a) sputtered without electric field, (b) spray pyrolysed without electric field and (c) spray pyrolysed with field of 32V/cm.

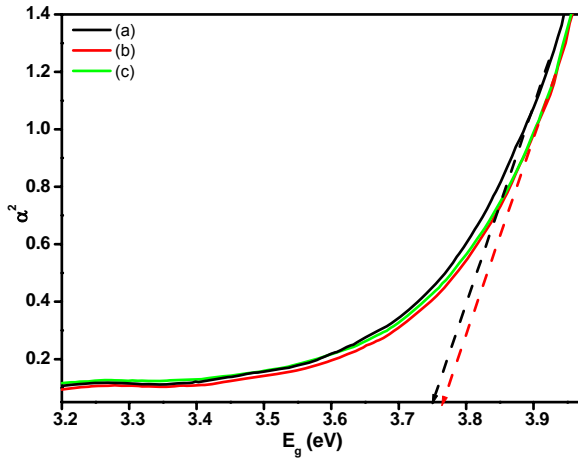


Fig.6. Plot of α^2 vs E_g of SnO₂ film (a) sputtered without electric field, (b) spray pyrolysed without electric field and (c) spray pyrolysed with field of 32V/cm.

From the transmission spectra, optical parameters like absorption coefficient (α), and optical band gap, E_g have been calculated. The absorption coefficient (α), have been determined by using the relation:

$$\alpha = \frac{\ln((1-R)/T)}{d} \quad (3)$$

where d is the thickness of the film in cm, T and R are the transmittance and reflectance of the film. The optical band gap energy of SnO₂ film was determined using the equation

$$\alpha h\nu = A(h\nu - E_g)^{1/2} \quad (4)$$

where A is a constant. To calculate the optical energy band gap value for a particular film α^2 vs. energy graph is plotted, as in figure 6, and by extrapolating the straight line, the intercept on the energy axis at $\alpha = 0$ is determined. The optical energy band gap is found to be nearly 3.75 eV.

4 CONCLUSIONS

Highly conducting and transparent ultrathin pristine SnO₂ films were fabricated on glass by electric field modified spray pyrolysis and DC reactive sputtering techniques. SnO₂ films grown by both the techniques were transparent in visible region with transparency of about 90% and band gap was found to be 3.75eV. The electrical conductivity of the films of 30 nm thickness was enhanced by more than one order i.e. from 3.8 to 14.7ohm⁻¹cm⁻¹ when deposited in the presence of electric field which is interpreted in terms of an increase the carrier density. So we can conclude that the electric field plays a pivotal role in controlling the physical properties of ultrathin films.

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